

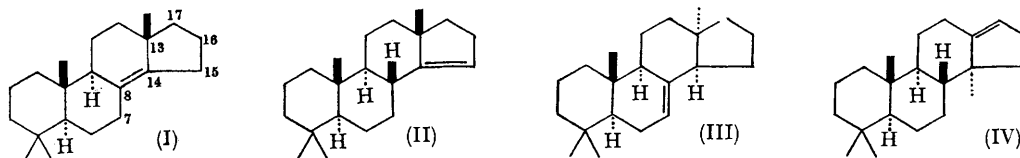
Methyl Migration in the Rearrangement of a 13-Isosteroid

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THE acid-catalysed rearrangement of steroidal olefins and dienes, unsaturated in rings B and C, to products unsaturated in ring D, is well known, and has found application in the synthesis of both di-^{1,2} and tri-terpenes.³ For example, 4,4-dimethyl-androst-8(14)-ene (I) is smoothly converted by

hydrogen chloride in the presence of palladium-charcoal catalyst to the Δ^{14} -isomer (II).¹ We report that this simple double-bond migration does not take place in the 13 α -steroid system, but that instead the 13 α -methyl group undergoes a 1,2-shift to C-14. Thus 4,4-dimethyl-13 α -androst-7-ene



(III) is converted under conditions similar to the above but more slowly to a mixture from which the 14 α -methyl-18-nor-derivative (IV) can be isolated in 51% yield. A second acid treatment of the residue after separation of (IV) yields further quantities of the rearranged olefin. Since (III) is itself readily available by photoepimerisation⁴ and Wolff-Kishner reduction of 4,4-dimethylandrosta-7-en-17-one, the reaction provides a convenient and stereospecific route to the novel 14 α -methyl-18-norsteroid system.

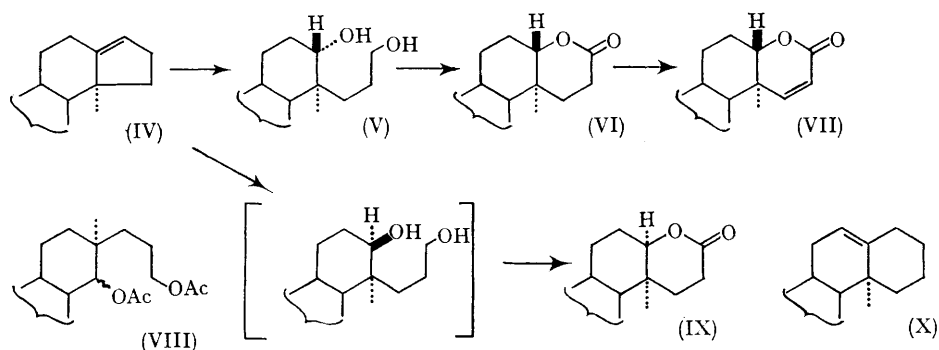
Structure (IV) is assigned to the rearrangement product on the following evidence. The double bond is trisubstituted (τ , 4.96 p.p.m.). Ozonolysis of (IV) and reduction of the ozonide with lithium aluminium hydride yielded a mixture of two epimeric diols, (*cf.* ref. 1), of which the major and more polar component (V) could be separated in 41% yield. The diacetate showed the C-13 proton as a quartet (τ 5.34 p.p.m., splittings 4 and 11 c./sec.), unchanged at 60 and 100 Mc./sec., in deuteriochloroform and in benzene, and over a temperature range of 20–100°. These observations are not consistent with structures such as (VIII), which would arise from cleavage of an un-rearranged Δ^{14} -olefin.

Lactone (IX) showed the C-13 proton as a poorly resolved triplet (τ 5.96 p.p.m., band width 6 c./sec.), thus confirming on the grounds of coupling constants⁵ and chemical shifts⁶ the stereochemical assignments at C-13 both in the lactones and their precursors. Furthermore, both lactones gave o.r.d. curves in agreement with prediction.⁷

Dehydrogenation⁸ of the lactone (VI) with dichlorodicyanobenzoquinone gave the dehydro-lactone (VII) (τ 4.14 and 3.01 p.p.m., one proton doublets, J 10 c./sec., C-15 and C-16 protons). The formation of a disubstituted olefin in this reaction eliminates the alternative structure (X) for the initial rearrangement product.

When the Δ^7 -olefin (III), tetradeuteriated at C-16 and C-17 (58% deuteration by mass spectrometry), was isomerised, nearly half the deuterium was lost in the product (IV). Thus both deuterium atoms at C-17 are largely eliminated during the isomerisation reaction, and the remaining deuterium is adjacent to the olefinic proton (τ 4.96 p.p.m., width at half height 3.0 c./sec., *cf.* undeuteriated (IV), width at half height 5.0 c./sec.).

Analysis of the rearrangement of (III) to (IV) is found in the conversion of euphenyl acetate to isoeuphenyl acetate,⁹ which may be a concerted

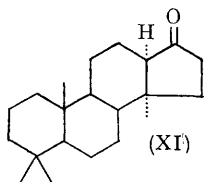


Manganese dioxide oxidation¹ of the diol (V) yielded the lactone (VI) (84%, τ 6.15 p.p.m., splittings 4 and 11 c./sec.). Similar oxidation of the mixed diols remaining after separation of (V) gave a mixture of lactones from which both (VI) and its C-13 epimer (IX) could be isolated.

process; it is probable that in our case the $\Delta^{8(14)}$ -isomer of (III) is an intermediate in the reaction. In the $\Delta^{8(14)}$ -isomer of (III), ring c occupies a boat or a half-boat conformation in which the C-13 methyl group takes on a quasi-axial orientation favourable for migration. Protonation at C-8 on

the β -face of the molecule then allows concerted rearrangement with formation of the stable all-*trans* steroid backbone in the product (IV).

The reaction may be contrasted with two examples in which Δ^7 -tetracyclic triterpenes are converted into Δ^{14} -olefins *without* migration of the adjacent 13α -methyl group. However, these reactions were effected by chromic acid oxidation¹⁰ or



by the action of boron trifluoride on the 7,8-epoxide,¹¹ and have apparently not been observed in simple proton-catalysed reactions. The present rearrangement has potential for the preparation of 18-nor-analogues of the known 14α -methyl-steroid hormones,¹² and also in the synthesis of triterpenes related to lanosterol. Thus in preliminary experiments, the olefin (IV) has been converted *via* the intermediate epoxide to the 17-ketone (XI). Ketone (XI) possesses the necessary function for the re-introduction of the 13β -methyl group and for the elaboration of a variety of side chains at C-17. The problem of retaining unsaturation in rings b and c and other aspects are under investigation.

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- ¹ P. Johnston, R. C. Sheppard, C. E. Stehr, and S. Turner, *J. Chem. Soc. (C)*, 1966, 1847.
- ² M. Fetizon and M. Golfier, *Bull. Soc. chim. France*, 1966, 850.
- ³ R. B. Woodward, A. A. Patchett, D. H. R. Barton, D. A. J. Ives, and R. B. Kelly, *J. Chem. Soc.*, 1957, 1131.
- ⁴ Cf. A. Butenandt and L. Poschmann, *Ber.*, 1944, **77**, 394; M. Fetizon and J.-C. Gramain, *Bull. Soc. chim. France*, 1967, 1003.
- ⁵ N. S. Bhacca and D. H. Williams, "Applications of N.M.R. Spectroscopy in Organic Chemistry", Holden-Day, San Francisco, 1964, pp. 81, 135.
- ⁶ Ref. 5, p. 47.
- ⁷ J. P. Jennings, W. Klyne, and P. M. Scopes, *J. Chem. Soc.*, 1965, 7211; W. Klyne, P. M. Scopes, R. C. Sheppard, and S. Turner, *J. Chem. Soc. (C)*, in the press.
- ⁸ B. Berkoz, L. Cuellar, R. Grezemkovsky, N. V. Avila, and A. D. Cross, *Proc. Chem. Soc.*, 1964, 215; see also ref. 1.
- ⁹ D. H. R. Barton, J. F. McGhie, M. K. Pradhan, and S. A. Knight, *J. Chem. Soc.*, 1955, 876.
- ¹⁰ W. Lawrie, W. Hamilton, F. S. Spring, and H. S. Watson, *J. Chem. Soc.*, 1956, 3272.
- ¹¹ G. P. Cotterrell, T. G. Halsall, and M. J. Wriglesworth, *Chem. Comm.*, 1967, 1121.
- ¹² G. R. Pettit and T. H. Brown, *J. Chem. Soc. (C)*, 1967, 2024 and references therein.